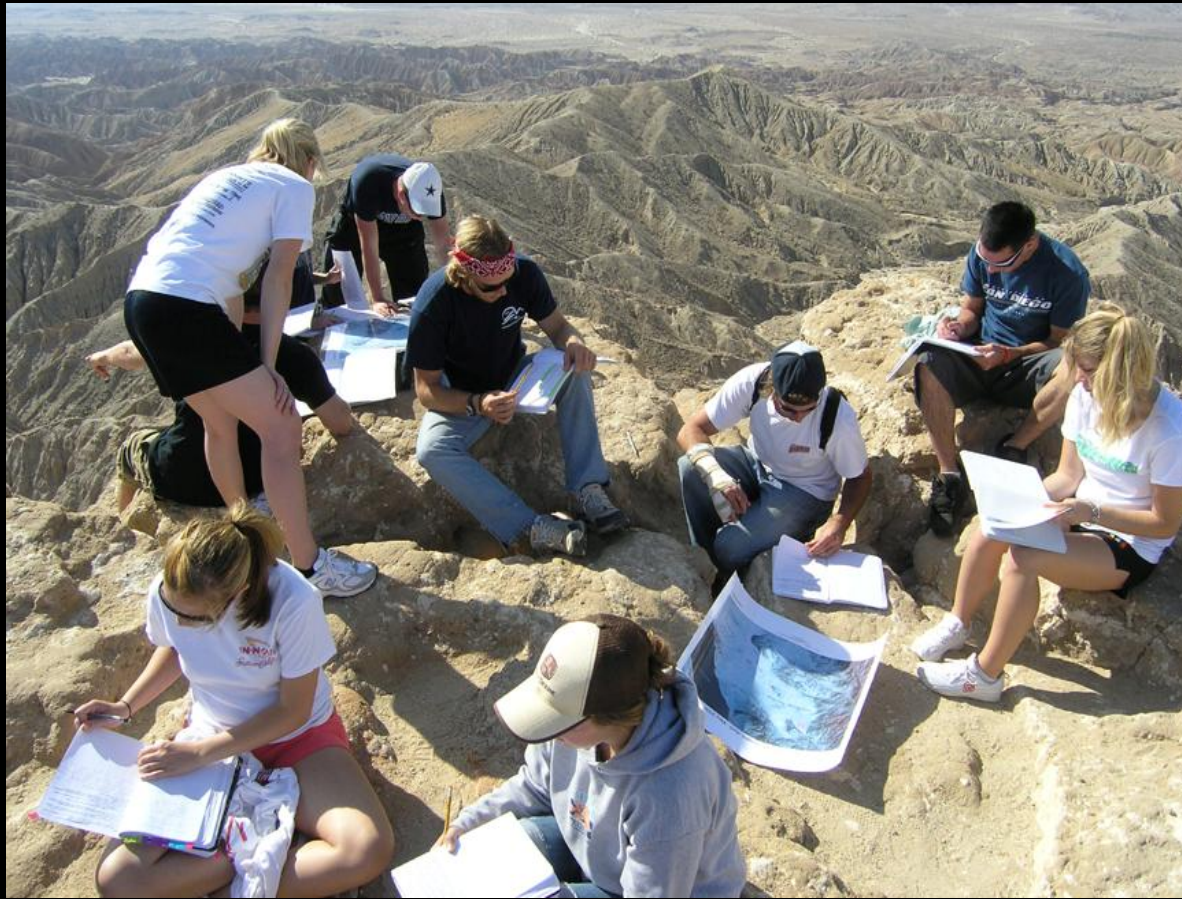
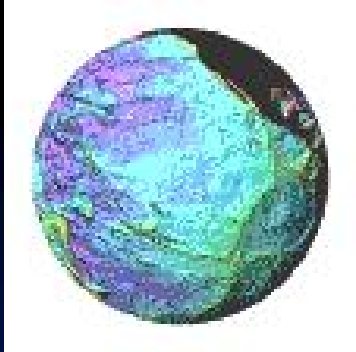


Physical Geology GEOL100



Instructor: Ray Rector



First Day Agenda

- Course Description
- Review of Course Syllabus
- Instructor Introduction
- Student Introductions
- Overview of Study of Geology
- Scientific Method

Course Description

- Online, self-study and discussion-based examination of geologic features and processes of the Earth's crust and mantle.
- Topics Include:
 - ★ Scientific Method
 - ★ Origins of Earth
 - ★ Plate Tectonics
 - ★ Minerals
 - ★ Rocks
 - ★ Volcanism
 - ★ Geologic Dating
 - ★ Earthquakes
 - ★ Structural Geology and Mountain Building
 - ★ Rivers and Groundwater
 - ★ Deserts and Glaciation
 - ★ Shoreline and Climate Change

Course Design



- Online-Based Format
- Course Activities Include:
 - ★ Reading assignments
 - ★ Online exercises
 - ★ Discussion forums
 - ★ Video instruction
 - ★ Instructor PowerPoints
 - ★ Instructor lecture note series
 - ★ Research writing assignments
 - ★ Quizzes and Exams

Course Syllabus

- Basic Logistics
- Course Objectives
- Important Enrollment Dates
- Instructor's Attendance Policy
- Blackboard Classroom Do's and Don'ts
- Grading
- Assignments and Testing
- Extra Credit
- [Classroom Website](#)
- Schedule of Study



www.geoscirocks.com

Miramar Online Lab Link



Professor Ray

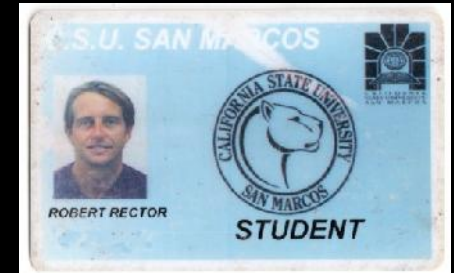
- Instructor's Academic Background
- Instructor's Passion for Geology
- Instructor's Role in Classroom
- Instructor's Teaching Philosophy

Who am I?

EARTH SCIENCE EDUCATION

California Single Subject Teaching Credential – Geosciences - California State University, San Marcos, CA

- 35 graduate-level semester units completed; GPA = 3.9
- Cross-Cultural Language and Academic Development
- Additional emphasis of technology in the classroom



Earth Science Doctoral Program – Volcanism and Tectonics University of California Riverside, Riverside, CA.

- 38 graduate-level semester units completed; GPA = 3.9
- Graduate Division Fellowship
- Mineralogical Society of America scholarship



Master of Science Degree – Igneous Petrology San Diego State University, San Diego, CA

- 35 graduate-level semester units completed; GPA=3.9
- Achievement Rewards for College Scientists Scholarship

Bachelor of Science Degree - Magna Cum Laude - Geology San Diego State University, San Diego, CA

- 172 semester units completed; GPA = 3.8
- Outstanding Senior Research Award--College of Sciences
- Outstanding Research Award—Department Of Geology



Engineering Undergraduate Program California State University, Northridge, CA

- Marine Engineering emphasis

TEACHING EARTH SCIENCE

Cuyamaca College, El Cajon, CA

- ❖ Oceanography Lecture

2013 - Present

University of San Diego, San Diego, CA

- ❖ Earth Science Laboratory

2007 - Present

MiraCosta College, Oceanside, CA

- ❖ Oceanography Lecture and Laboratory
- ❖ Online Geology

2004 - Present

San Diego Miramar College, San Diego, CA

- ❖ Geology Laboratory
- ❖ Online Oceanography Lecture

2003 - Present

San Diego Mesa College, San Diego, CA

- ❖ Online Geology Lecture
- ❖ Geology Laboratory

2002 - Present

University of California Riverside, Riverside, CA

- ❖ General geology, Historical geology, Mineralogy, Optical mineralogy, Igneous petrology, and Metamorphic petrology

1994-1997

San Diego State University, San Diego, CA

- ❖ General geology laboratory
- ❖ Advanced field geology course in Baja, Mexico.

1991-1993

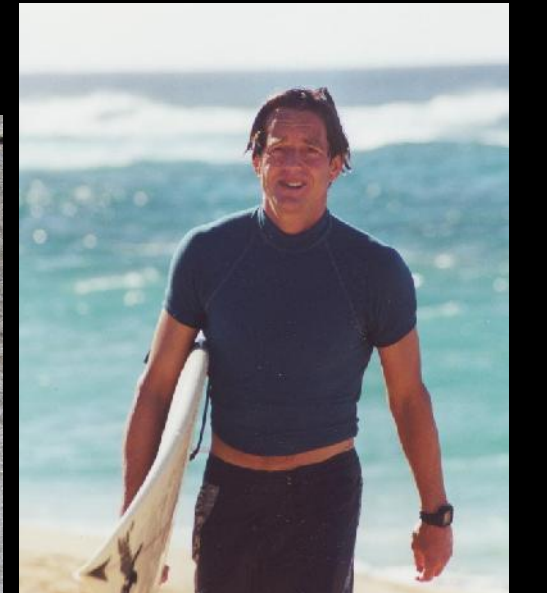
Professor's Interests



Travel to Cool Places, Adventure, Hanging Out, and Partying with Fun and Interesting Friends



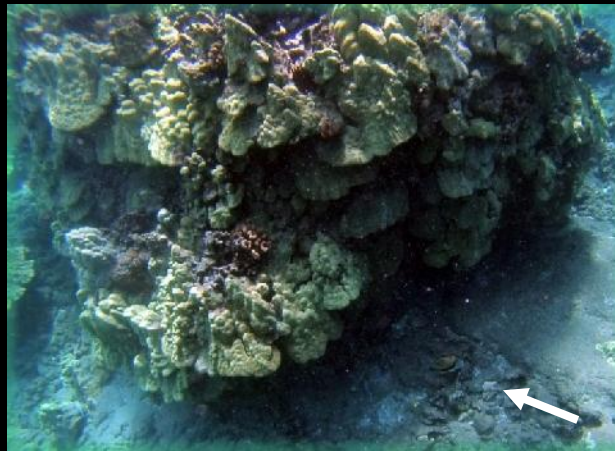
Outdoor Sports



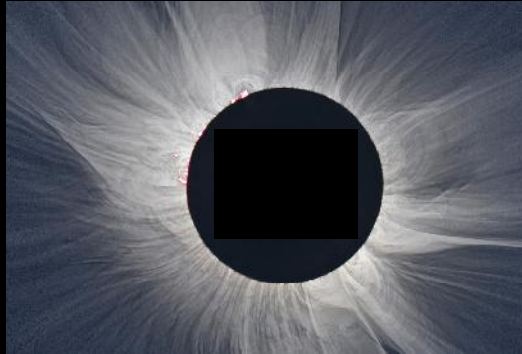
Last Summer's Big Adventure – The Big Island



*Aloha from
Hawaii!*



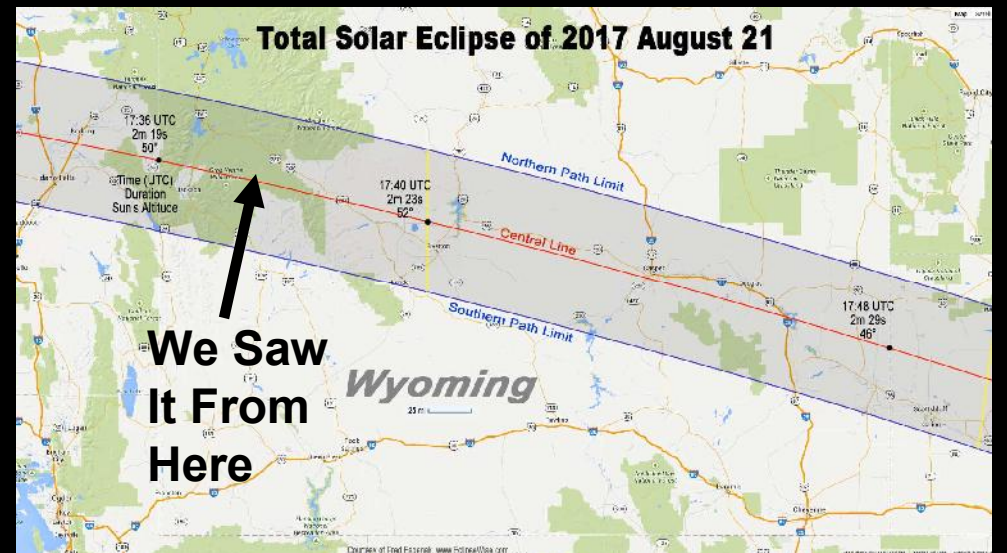
Last Summer's Adventure – Total Eclipse



Howdy from



Wyoming!





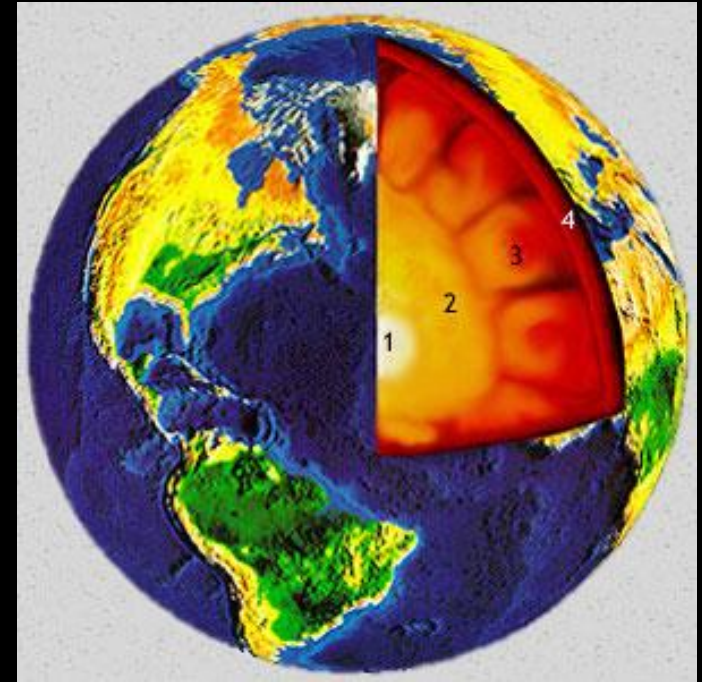
Who are You?

- Your Name
- Academic Focus
- Personal Interests
- Why take Geology Course?



Wishing Everyone a Great Fall Semester

Geology of Planet Earth



Geology of San Diego County



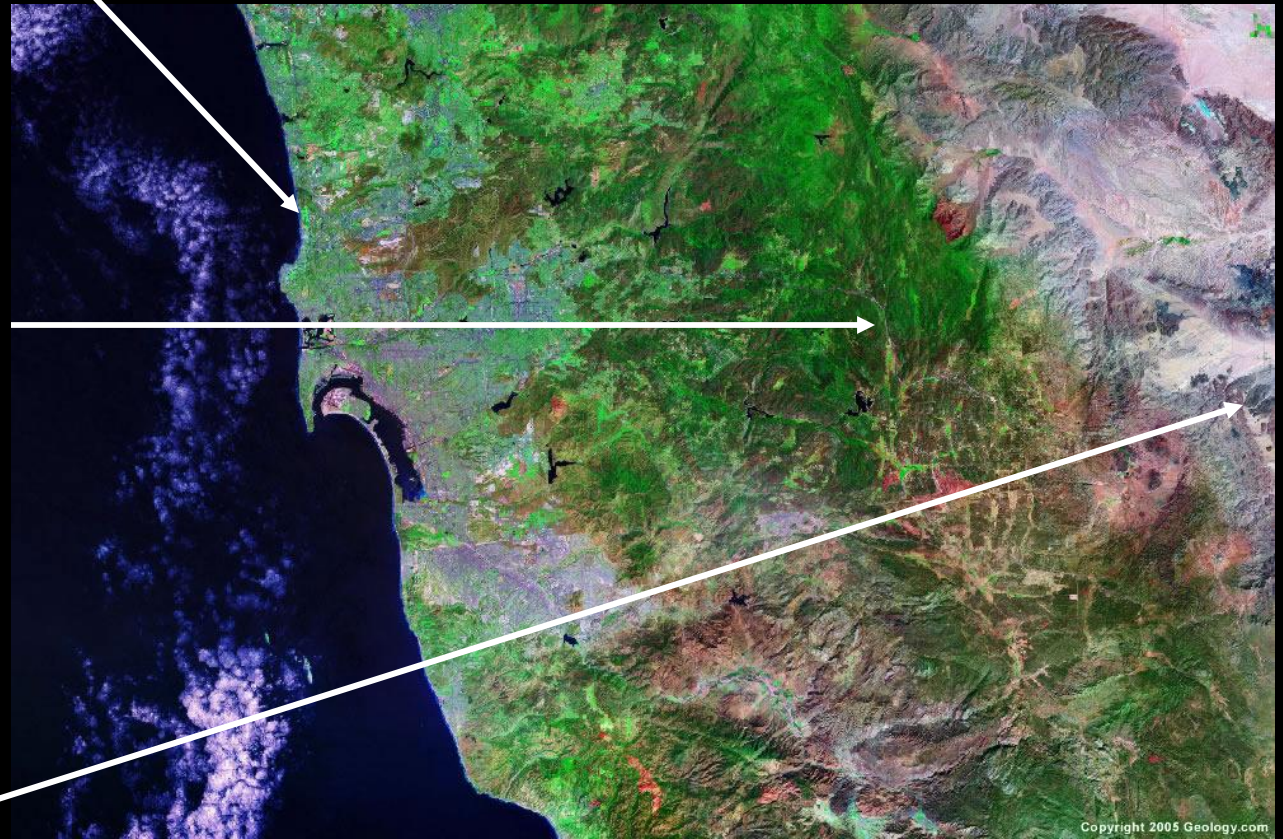
The Coastal Geology



Backcountry Geology



Desert Geology

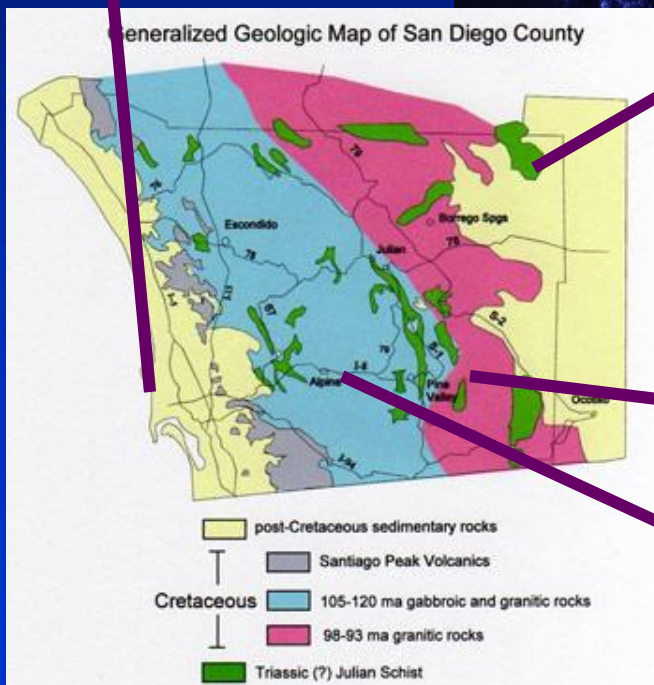


Geology of San Diego

Metamorphic



Sedimentary



Igneous

Local Natural Hazards



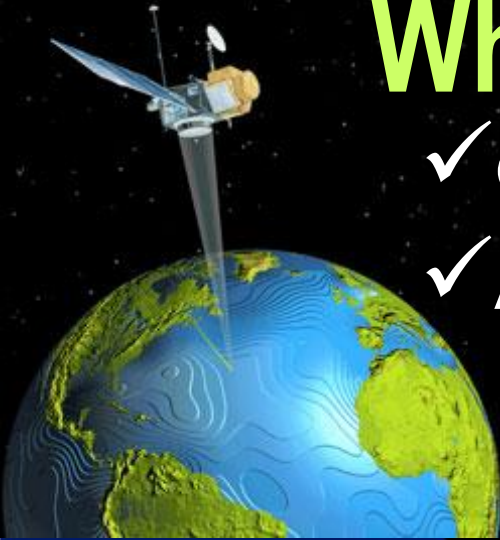
Homes on cliffs of Pacifica, CA

Question:

What other geologic hazards do we face in San Diego?

What is Geology?

- ✓ Geology is the scientific study of the Earth
- ✓ An interdisciplinary science



Hydrologic Studies



Volcanic Studies



Seismic Studies



Marine Studies



GEOLOGY -- an Interdisciplinary Science

Geology integrates many different types of geosciences

- **Mineralogy and Petrology** - the study of minerals and rocks
- **Marine geology** - the study of Earth's ocean bottom
- **Geochemistry**- study of chemical nature of rocks, minerals and fluids
- **Hydrology** - study of rivers, groundwater, flooding, dams
- **Volcanology** – study of the nature and distribution of volcanoes
- **Engineering geology**- design and construction of structures
- **Structural geology** - form and development of geologic structures
- **Geophysics** – study of forces and mechanisms of geologic phenomena
- **Environmental geology** – study of geological resources and pollution
- **Petroleum geology** – Locate, assess, and extract oil and natural gas

What Do Geologists Do?

Answer: ...they *do* earth science.



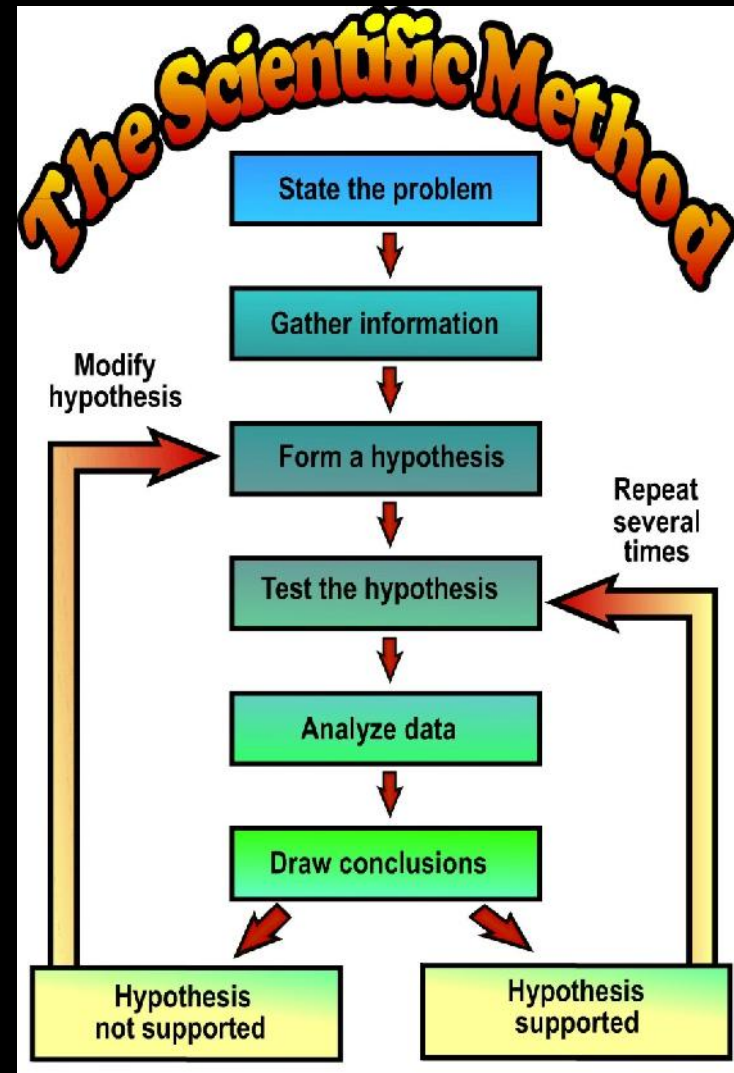
Science defined: The investigation and acquisition of useful, reliable knowledge of earth's crust that is based on empirical observations (physical evidence).

- ✓ Earth scientists use a powerful way of thinking, that is rational, logical, and organized, called ***scientific thinking***.
- ✓ Intelligence, imagination, creativity, inspiration, and luck are other important attributes of scientific study.
- ✓ Earth scientists use a powerful approach to inquiry called the **scientific method**.
- ✓ Central to science is **community and peer review**.

THE SCIENTIFIC METHOD

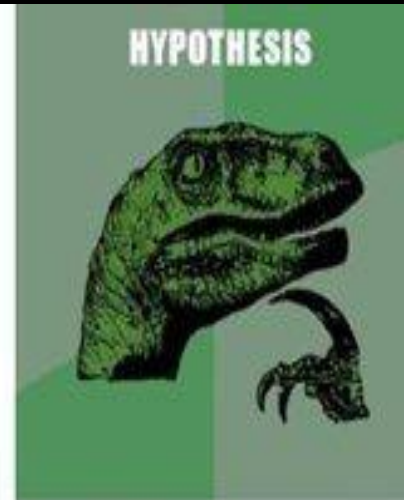
The Basic Components

- ✓ *Empirical Observations*
- ✓ *Questions / Problems*
- ✓ *Hypotheses / Models*
- ✓ *Predictions*
- ✓ *Tests / Experiments*
- ✓ *Analysis of Results*
- ✓ *Draw Conclusions*
- ✓ *Reevaluate Hypothesis*



Note: The scientific method is NOT a recipe – it's a process ²³

Today's Ocean Lab: *Investigation and Application of the Scientific Method*



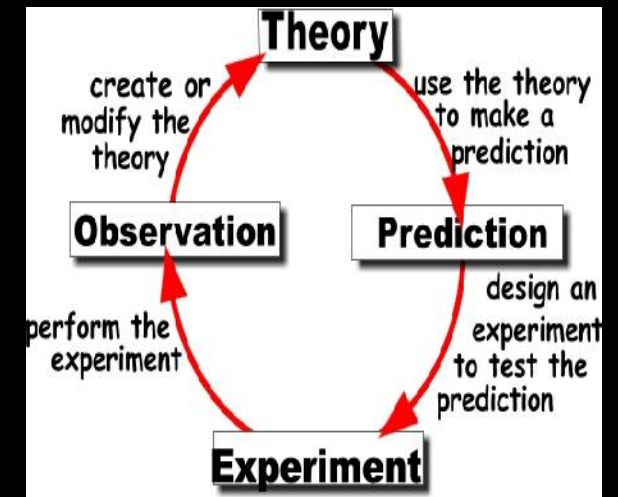
Rationalism
(Logic & Reasoning)



Empiricism
(Experience & Observation)



Science



Investigation and Application of the Scientific Method

Scientific Method

■ O

OBSERVATION



■ H

HYPOTHESIS



■ T

TEST



Row	Color	Elements in Spectrum	Class	Other Observations
1				
2				
3				
4				
5				
6				
7				
8				
9				
10				

■ A

ANALYZE DATA

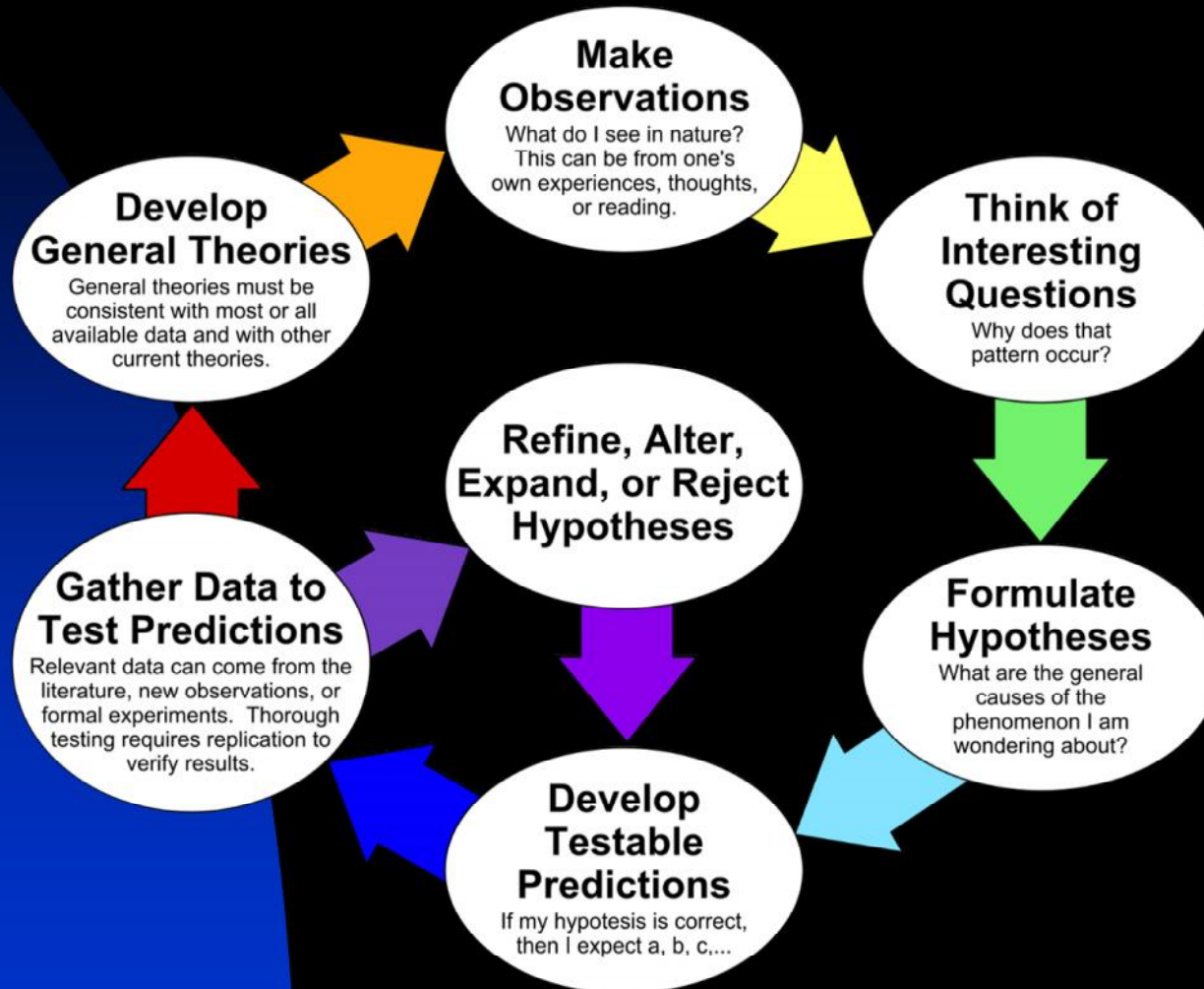


■ C

CONCLUSION



Scientific Method is an Ongoing Process



Scientific *Observations*

Making Observations

- ▶ There are two different types of observations - **quaLitative** observations and **quaNtitative** observations.

Quantitative Units of Measurement

US Standard System of Units

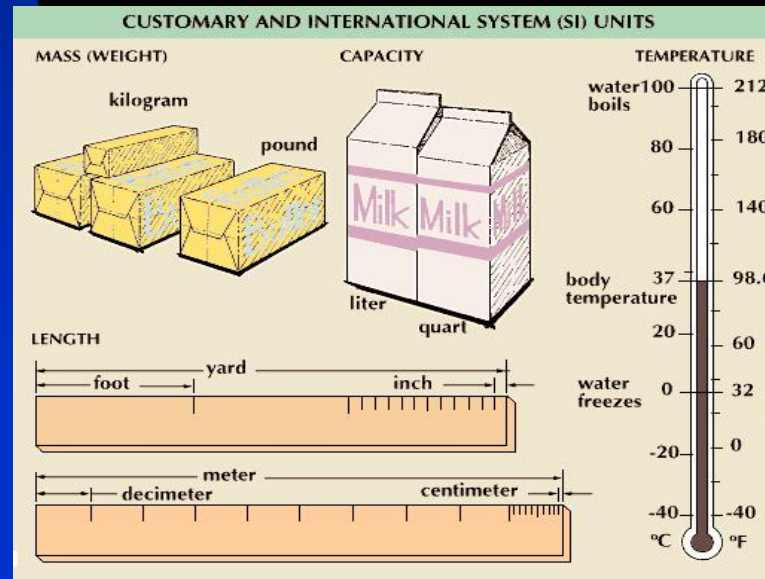
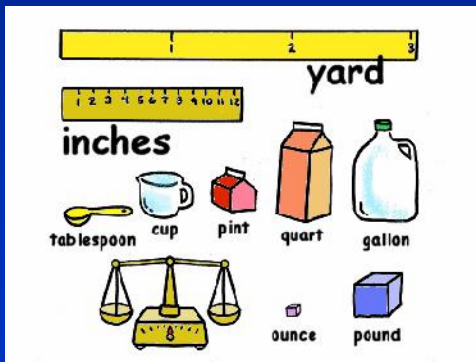
inch/foot
square foot
ounce/gallon
ounce/pound
second
Fahrenheit

Measurable Physical Quantities

- 1) Distance -
- 2) Area -
- 3) Volume -
- 4) Mass -
- 5) Time -
- 6) Temperature -

International Metric System of Units

centimeter/meter
square meter
milliliter/liter
gram/kilogram
second
Kelvin/Celsius



Physical quantity measured	Base unit	SI abbreviation
	mole	mol
	meter	m
	kilogram	kg
	second	s
	kelvin	K
	ampere	A
	candela	cd

International Metric Units

Quantity measured	Unit	Symbol	Relationship
Length, width, distance, thickness, girth, etc.	millimeter	mm	10 mm = 1 cm
	centimeter	cm	100 cm = 1 m
	meter	m	
	kilometer	km	1 km = 1000 m
Mass ("weight")*	milligram	mg	1000 mg = 1 g
	gram	g	
	kilogram	kg	1 kg = 1000 g
	metric ton	t	1 t = 1000 kg
Time	second	s	
Temperature	degree Celsius	° C	
Area	square meter	m ²	
	hectare	ha	1 ha = 10 000 m ²
	square kilometer	km ²	1 km ² = 100 ha
Volume	milliliter	mL	1000 mL = 1 L
	cubic centimeter	cm ³	1 cm ³ = 1 mL
	liter	L	1000 L = 1 m ³
	cubic meter	m ³	
Speed, velocity	meter per second	m/s	
	kilometer per hour	km/h	1 km/h = 0.278 m/s

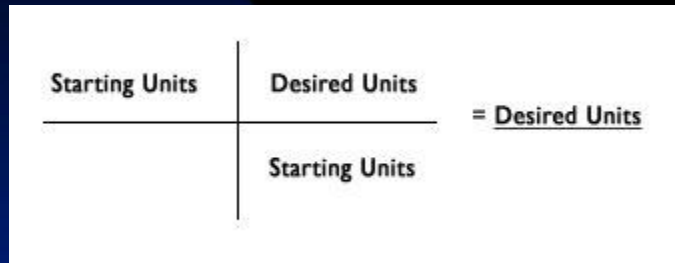
Metric Unit Prefixes

Prefix	Symbol	Factor	Numerically	Name
giga	G	10^9	1 000 000 000	billion**
mega	M	10^6	1 000 000	million
kilo	k	10^3	1 000	thousand
centi	c	10^{-2}	0.01	hundredth
milli	m	10^{-3}	0.001	thousandth
micro	μ	10^{-6}	0.000 001	millionth
nano	n	10^{-9}	0.000 000 001	billionth**

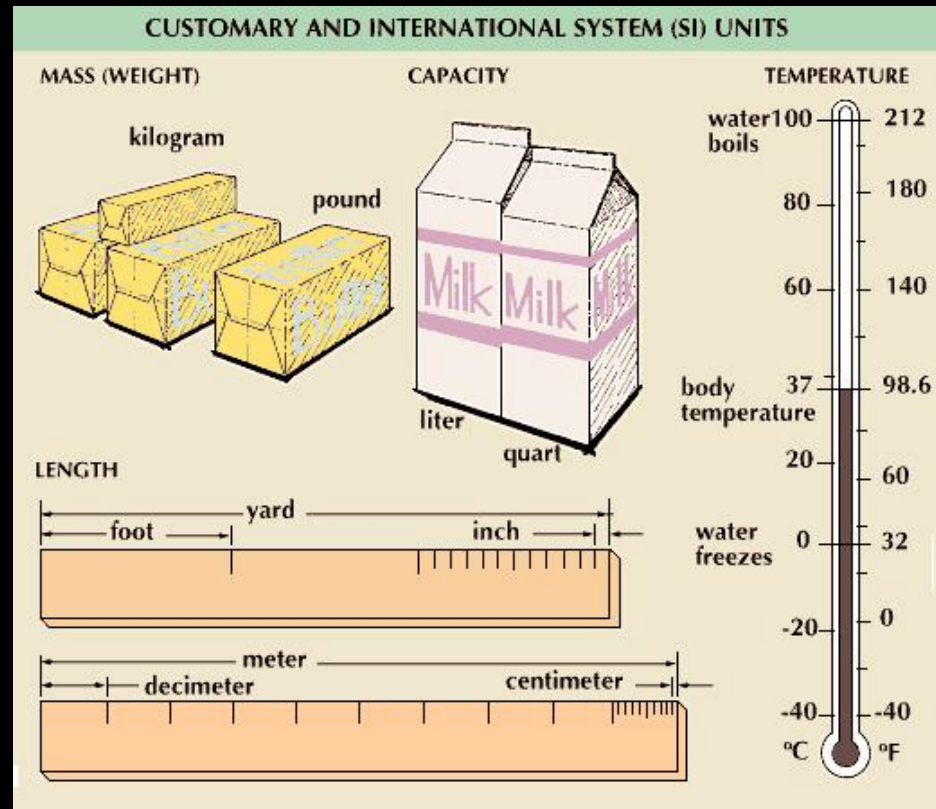
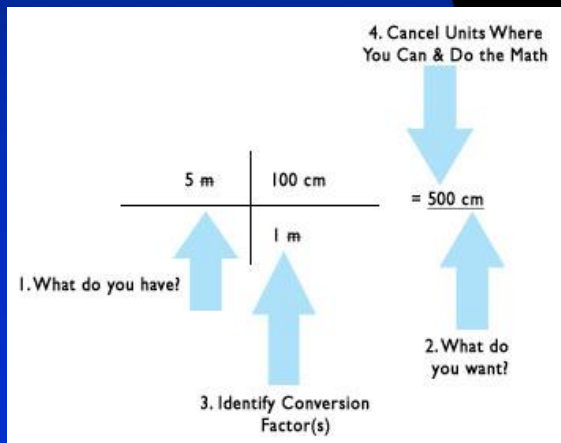


Converting Units of Measurement

Setting Up the Problem:



Example:
Convert 15 m to ? cm



Converting Units

Make sure to:

- 1) Find the proper conversion factor for the two units
- 2) Set up the equation with all numeric values having a unit symbol
- 3) Do the conversion making sure that the old unit cancels

APPROXIMATE CONVERSIONS FROM ENGLISH UNITS TO SI UNITS				
SYMBOL	WHEN YOU KNOW	MULTIPLY BY (CF)	TO FIND	SYMBOL
LENGTH				
in	inches	25.4	millimeters	mm
ft	feet	0.305	meters	m
yd	yards	0.914	meters	m
mi	miles	1.61	kilometers	km
AREA				
in ²	square inches	645.2	square millimeters	mm ²
ft ²	square feet	0.093	square meters	m ²
yd ²	square yard	0.836	square meters	m ²
ac	acres	0.405	hectares	ha
mi ²	square miles	2.59	square kilometers	km ²
VOLUME				
fl oz	fluid ounces	29.57	milliliters	mL
gal	gallons	3.785	liters	L
ft ³	cubic feet	0.028	cubic meters	m ³
yd ³	cubic yards	0.765	cubic meters	m ³
NOTE: volumes greater than 1000 L shall be shown in m ³				
MASS				
oz	ounces	28.35	grams	g
lb	pounds	0.454	kilograms	kg
T	short tons (2000 lb)	0.907	megagrams (or "metric ton")	Mg (or "t")
TEMPERATURE (exact degrees)				
°F	Fahrenheit	5 (F-32) ÷ 9	Celsius	°C

Metric Conversion Chart and Table

Length

1 centimeter (cm)	=	10 millimeters (mm)
1 inch	=	2.54 centimeters (cm)
1 foot	=	0.305 meters (m)
1 foot	=	12 inches
1 yard	=	3 feet
1 meter (m)	=	100 centimeters (cm)
1 meter (m)	≈	3.281 feet
1 furlong	=	660 feet
1 kilometer (km)	=	1000 meters (m)
1 kilometer (km)	≈	0.62137119 miles
1 mile	=	5280 ft
1 mile	=	1.61 kilometers (km)
1 nautical mile	=	1.85 kilometers (km)

Area

1 square foot	=	144 square inches
1 square foot	=	929.03 square centimeters
1 square yard	=	9 square feet
1 square meter	≈	10.76104 square feet
1 acre	=	43,560 square feet
1 hectare	=	10,000 square meters
1 hectare	≈	2.47 acres
1 square kilometer	=	100 hectares
1 square mile	≈	2.59 square kilometers
1 square mile	=	640 acres

Speed

1 mile per hour (mph)	≈	1.467 feet per second (fps)
1 mile per hour (mph)	=	1.61 kilometers per hour
1 knot	≈	1.15 miles per hour
1 foot per second	≈	0.68 miles per hour (mph)
1 kilometer per hour	≈	0.62 miles per hour (mph)

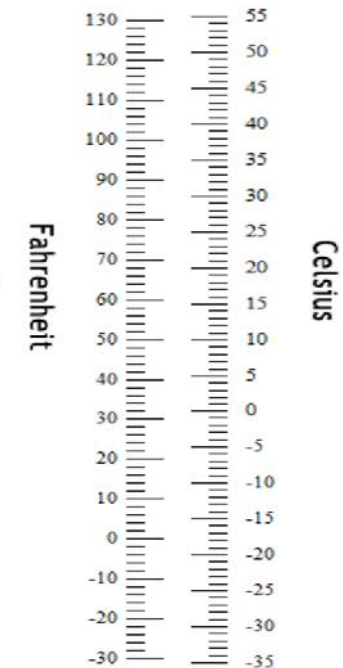
Volume

1 US tablespoon	=	3 US teaspoons
1 US fluid ounce	≈	29.57 milliliters (ml)
1 US cup	=	16 US tablespoons
1 US cup	=	8 US fluid ounces
1 US pint	=	2 US cups
1 US pint	=	16 US fluid ounces
1 liter (l)	≈	33.81 US fluid ounces
1 liter (l)	=	1000 milliliters (ml)
1 US quart	=	2 US pints
1 US gallon	=	4 US quarts
1 US gallon	=	3.785 liters

Weight

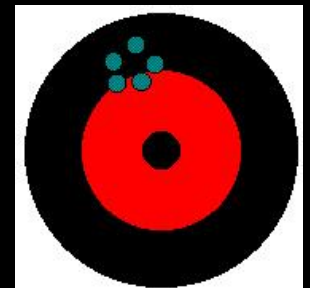
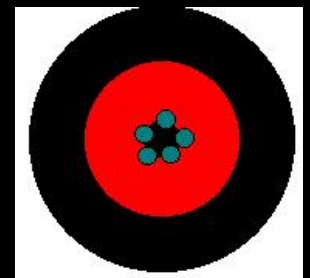
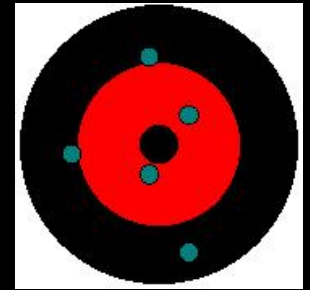
1 milligram (mg)	=	0.001 grams (g)
1 gram (g)	=	0.001 kilograms (kg)
1 gram (g)	≈	0.035 ounces
1 ounce	=	28.3 grams (g)
1 ounce	=	0.0625 pounds
1 pound (lb)	=	16 ounces
1 pound (lb)	=	0.45 kilograms (kg)
1 kilogram (kg)	=	1000 grams
1 kilogram (kg)	≈	35.27 ounces
1 kilogram (kg)	≈	2.2 pounds (lb)
1 stone	=	14 pounds
1 short ton	=	2000 pounds
1 metric ton	=	1000 kilograms (kg)

Temperature



Accuracy, Precision and Uncertainty in Measurement

- 1) **Accuracy** of the measurement refers to how close the measured value is to the true or accepted value.
- 2) **Precision** refers to how close together a group of measurements actually are to each other.
- 3) Accuracy can be determined by only one measurement, while precision can only be determined with multiple measurements.
- 4) Precision has nothing to do with the true or accepted value of a measurement, so it is quite possible to be very precise and totally inaccurate.
- 5) When precision is high and accuracy is low, the fault can lie with the instrument.



Significant Digits or Figures

Rules For Significant Digits

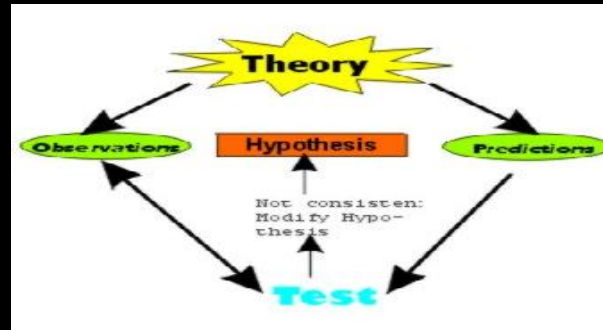
1. Digits from 1-9 are always significant.
2. Zeros between two other significant digits are always significant
3. One or more additional zeros to the right of both the decimal place and another significant digit are significant.
4. Zeros used solely for spacing the decimal point (placeholders) are not significant.

EXAMPLES	# OF SIG. DIG.	COMMENT
453 g	3	All non-zero digits are always significant.
5057 L	4	Zeros between 2 sig. dig. are significant.
5.00 ml	3	Additional zeros to the right of decimal and a sig. dig. are significant.
0.007 km	1	Placeholders are not sig.

Hypotheses and Scientific *Testing*



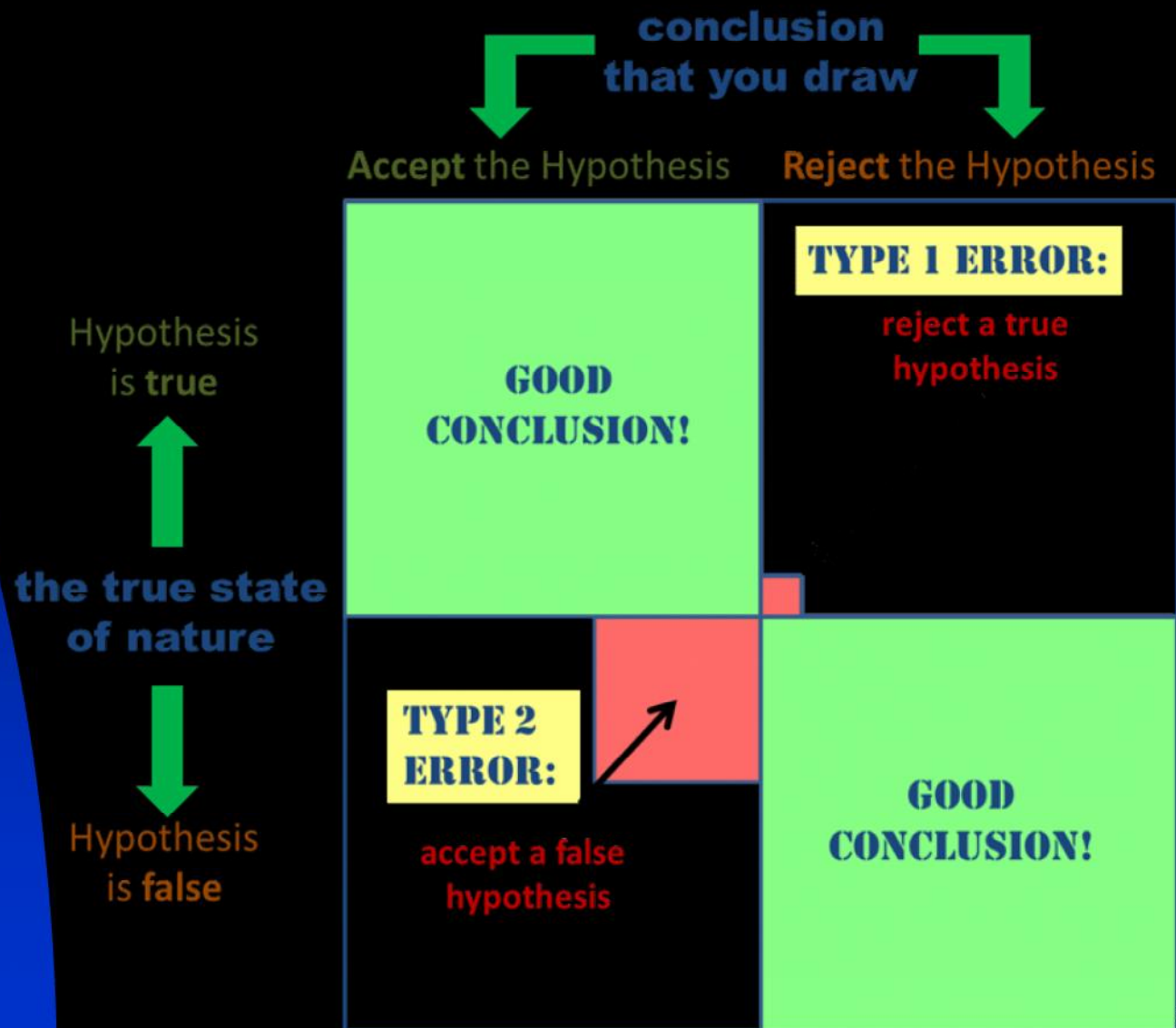
Observations and Predictions



Testing Hypotheses and Theories

- 1) A hypothesis is a simple explanation, model, or prediction of nature that requires testing (attempt to falsify or confirm).
- 2) Hypotheses are based on empirical physical evidence (data).
- 3) Hypotheses must be falsifiable (testable/predictable).
- 4) Hypotheses can never be proven as an absolute fact.
- 5) Hypotheses are always open to elimination or modification.
- 6) A theory is a broad, elegant, unifying explanation of a set of otherwise unconnected natural phenomena.
- 7) A theory is established by the interconnection (framework) of well-tested and confirmed hypotheses that are, in turn, supported by an enormous amount of physical evidence.

Testing Your Hypothesis



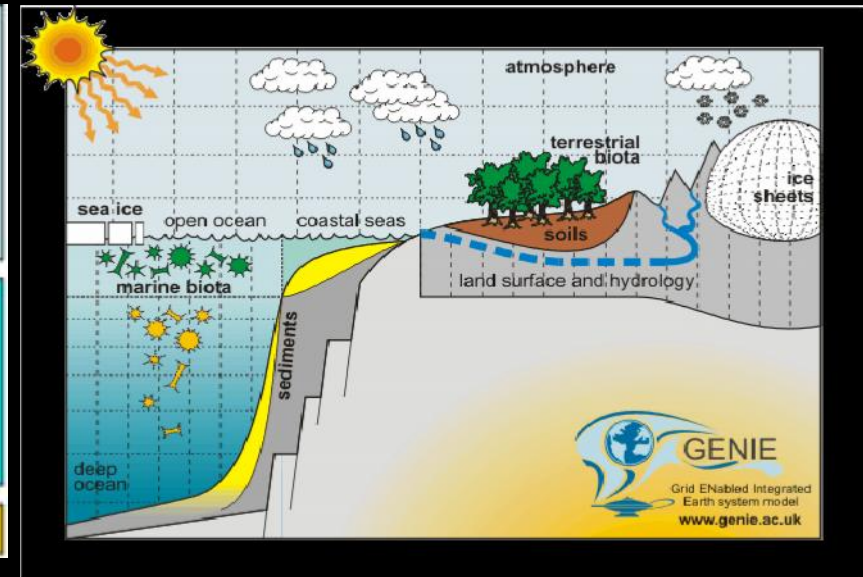
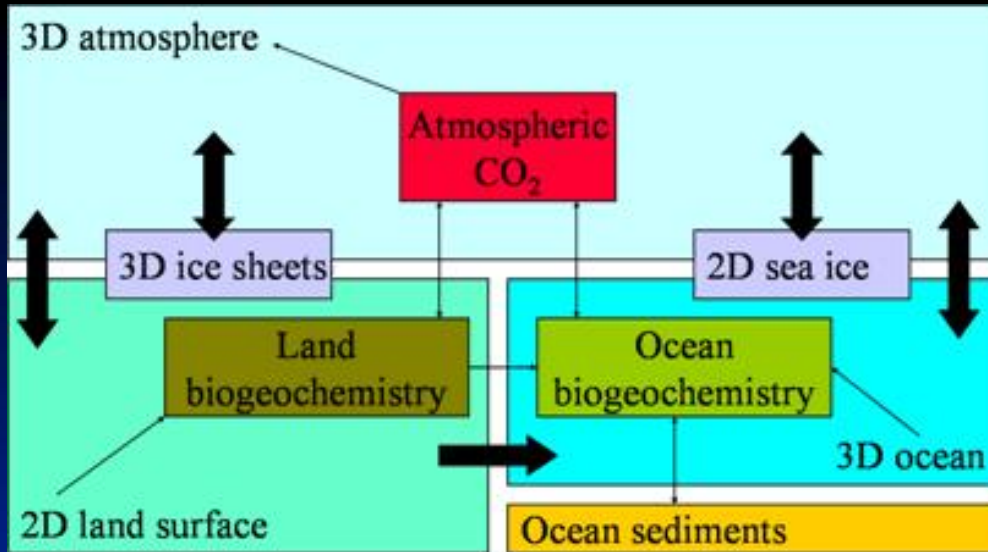
Scientific *Predictions*

Prediction

- A statement of what may happen in the future based on observations, data, experience or scientific reason



Scientific Modeling and Predicting



Purpose of Modeling: Understand and predict how parts of the Earth operate and interact with each other

- Start simple and get more complicated over time
- Add more and more parameters over time
- Test computer models with real historic data
- Develop and refine models to predict future scenarios

Application of the Scientific Method

Glitter Lamp Inquiry

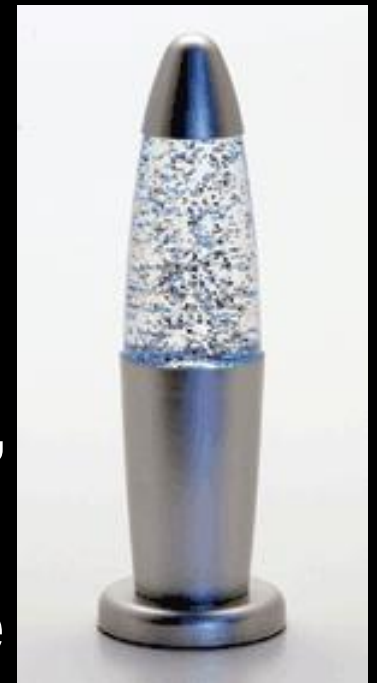
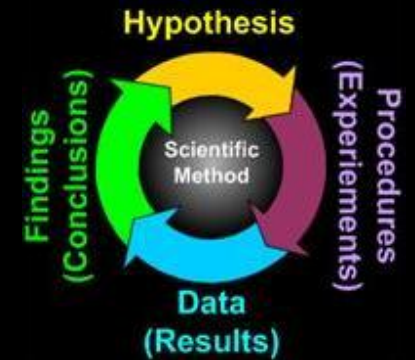
Purpose:

Use the scientific method to gain a better understanding of how a glitter lamp works as a dynamic system

Procedure:

Make good observations, explanations, predictions, and tests on the lamp.

Focus on the dynamic properties of the lamp.



Lava Lamp as a Model for Convection

Convection Process

- ✓ Fluid material at top of lamp is cooler than material at the bottom.
- ✓ Hotter material is less dense than cooler material
- ✓ Less dense fluid rises while more dense fluid sinks
- ✓ Heat and gravity drive the system
- ✓ Earth's atmosphere, ocean, mantle and core undergo convection

